

STEROIDS XXVI(1)

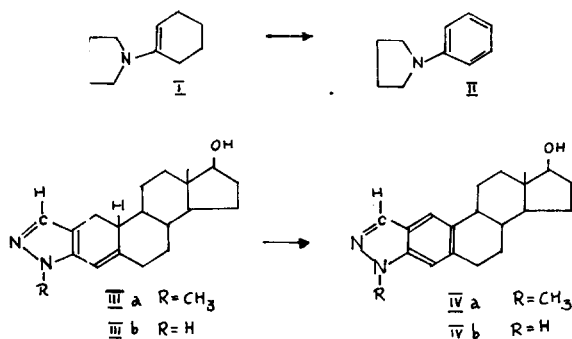
AROMATIZATION REACTIONS OF SOME 19-NOR-ANDROSTANE DERIVATIVES

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Recently, G. Bianchetti(2), refluxing cyclohexanone enamines in dioxane, noted the disproportionation of the same to N-cycloalkyl-anilines and N-cycloalkyl-cyclohexylamines. In addition, when the reaction was conducted in the presence of Pd/C and hydrogen acceptors (such as ethyl cinammate or stilbene) he obtained quantitative aromatization of cyclohexanone enamines (I for instance) and in the meantime quantitative formation of ethyl phenylpropionate or 1,2-diphenyl-ethane.

Some [2,3-d]-pyrazoles of  $\Delta^4$ -19-nor-androstene being synthesized by us, we considered the heterocyclic nitrogen in position 3, structurally similar to the enaminic nitrogen of Bianchetti's pyrrolydyl-cyclohexanone(I)(2) and thus we assumed that the above mentioned compound might undergo aromatization of the steroidal A-ring to estratriene through a similar mechanism.



Accordingly, we allowed to react [2,3-d]-1'-methyl-pyrazole-estra-4-ene-17 $\beta$ -ol (IIIa)\*, m.p. 209-211°,  $[\alpha]_D = +12^\circ$  (CHCl<sub>3</sub>),  $\lambda_{\max} 276 \text{ m}\mu$  \*\* (lg  $\epsilon = 4.00$ ) under reflux in dioxane with two moles of ethyl maleate and 10% Pd/C as hydrogen transfer. After 24 hours we isolated from the reaction mixture the [2,3-d]-1'-methyl-pyrazole-estra-1,3,5(10)-triene-17 $\beta$ -ol (IVa), m.p. 191-193°,  $[\alpha]_D = +8^\circ$  (CHCl<sub>3</sub>);  $\lambda_{\max} 257, 263, 272, 293, 299, 304, 312 \text{ m}\mu$  (lg  $\epsilon = 3.81; 3.85; 3.81; 3.73; 4.74; 3.72; 3.64$ ); I.R. 1639, 1504 and 883 cm<sup>-1</sup> (aromatic absorption).

Similarly, starting from IIIb, m.p. 242-245°,  $[\alpha]_D = +27^\circ$  (pyridine);  $\lambda_{\max} 260 \text{ m}\mu$  (lg  $\epsilon = 3.95$ ) we obtained IVb, m.p. 170-173°,  $\lambda_{\max} 260, 266, 289, 305 \text{ m}\mu$  (lg  $\epsilon = 3.98; 3.96; 3.86; 3.69$ ); I.R. 1639, 1504 and 883 cm<sup>-1</sup> (aromatic absorption).

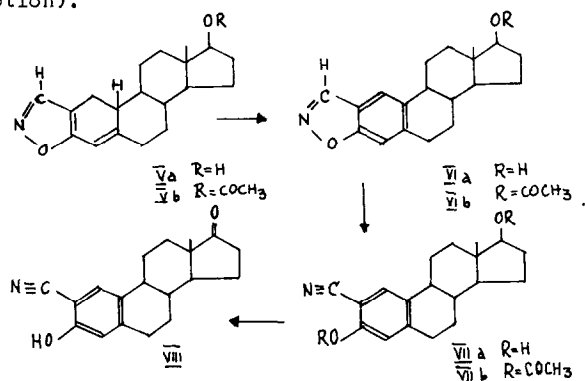
The U.V. spectra of the aromatization products IVa,b are in good agreement with those of the simpler benzopyrazoles (4).

The reaction, which was expected to be only limited to enamines, proved much more employable and not closely connected with the presence of an enaminic or enaminic-like structure.

\* - III and IIIb were prepared according to the usual procedure (3), starting from 2-hydroxymethylene-estra-4-ene-17 $\beta$ -ol-3-one (1).

\*\* - All I.R. spectra were executed in nujol suspension, while the U.V. spectra were determined in ethanolic solution.

Under similar experimental conditions (ethyl maleate, 10% Pd/C in dioxane under reflux), but starting from [2,3-d]-isoxazole (Va)(1), through the probable intermediate benzoisoxazole (VIa) we obtained 2-cyano-estra-1,3,5(10)-triene-3,17 $\beta$ -diol (VIIa)(1), m.p. 336-338°,  $[\alpha]_D^{25} = +41^\circ$  (pyridine),  $\lambda_{\max}^{236,305} \text{ m}\mu$  ( $\lg \epsilon = 4.04; 3.57$ );  $\lambda_{\max}^{\text{EtOH-NaOH}} 223, 246, 337 \text{ m}\mu$  ( $\lg \epsilon = 4.39; 4.00; 3.59$ ); I.R. 3425, 3125  $\text{cm}^{-1}$  (-OH); 2227  $\text{cm}^{-1}$  (-C $\equiv$ N); 1616, 1506, 894, 873  $\text{cm}^{-1}$  (aromatic absorption), which was acetylated to give VIIb, m.p. 149-150°,  $[\alpha]_D^{25} = +45^\circ$  (CHCl<sub>3</sub>);  $\lambda_{\max}^{236, 278, 288} \text{ m}\mu$  ( $\lg \epsilon = 4.05; 3.27; 3.25$ ). I.R. 2230  $\text{cm}^{-1}$  (-C $\equiv$ N); 1783  $\text{cm}^{-1}$  (C-3 acetate); 1739  $\text{cm}^{-1}$  (C-17 acetate); 1610, 1575, 1500, 897  $\text{cm}^{-1}$  (aromatic absorption).



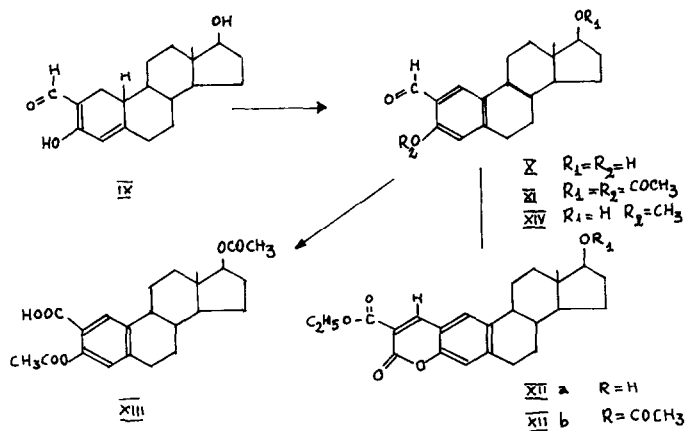
Oxidation of VIIa, conducted with Jones' reagent(5) afforded 2-cyano-estra-1,3,5(10)-triene-3-ol-17-one (VIII), m.p. 264-266°,  $\lambda_{\max}^{236,304} \text{ m}\mu$  ( $\lg \epsilon = 4.04; 3.60$ ); I.R. 3322  $\text{cm}^{-1}$  (OH-phenolic); 2232  $\text{cm}^{-1}$  (-C $\equiv$ N); 1730  $\text{cm}^{-1}$  (17-ketone); 1616, 1592, 1511, 883  $\text{cm}^{-1}$  (aromatic absorption).

The facility with which these aromatization reactions occurred, prompted us to verify the behaviour of some  $\Delta^4$ -estrene derivatives in the presence of 2,3-dichloro-5,6-dicyano-1,4-benzoquinone, which is widely employed in steroid chemistry for dehydrogenation.

Working in dioxane under conditions similar to those employed by J.A. Edwards et al. (6) for the preparation of 2-formyl-3-keto-steroids, or in benzene under reflux starting from 2-hydroxymethylene-estra-4-ene-17 $\beta$ -ol-3-one (IX)(4) we obtained in quantitative yields 2-formyl-estra-1,3,5(10)-triene-3,17 $\beta$ -diol(X), m.p. 232-234 $^{\circ}$ ,  $[\alpha]_D = +88^{\circ}$  (dioxane);  $\lambda_{max} 226.5, 267.5, 337.5 \mu$  ( $lg \epsilon = 4.19; 4.16; 3.54$ );  $\lambda_{max}^{EtOH-NaOH} 239, 275, 289 \mu$  ( $lg \epsilon = 4.25; 4.02; 3.77$ ); I.R. 3390, 3125  $cm^{-1}$  (-OH); 1664  $cm^{-1}$ , 1618  $cm^{-1}$  (formyl chelated form); 877  $cm^{-1}$  (aromatic absorption).

Its diacetate (XI) shows m.p. 151-152 $^{\circ}$ ,  $[\alpha]_D = +53^{\circ}$  (CHCl<sub>3</sub>),  $\lambda_{max} 261$  and 295.5  $\mu$  ( $lg \epsilon = 4.05; 3.36$ ); I.R. no OH; 1783  $cm^{-1}$  (C-3 acetate); 1745  $cm^{-1}$  (C-17 acetate); 1692  $cm^{-1}$  (C-2 formyl), 1610, 1575, 1563, 1550, 892  $cm^{-1}$  (aromatic absorption).

The 2-formylestradiol provides the typical reactions of aromatic o-hydroxy-aldehydes: yellow colour, when treated with alkaline hydroxide and violet colour, when reacted with FeCl<sub>3</sub>. In the presence of piperidine X reacts with ethyl malonate to give estra-1,3,5(10)-triene-17 $\beta$ -ol-[2,3-d]-2'-carboethoxy-2',3'-dihydropyrane-2'-one (XIIa), m.p. 188-190 $^{\circ}$ ;  $[\alpha]_D = +83^{\circ}$  (CHCl<sub>3</sub>),  $\lambda_{max} 307, 344 \mu$  ( $lg \epsilon = 4.16; 3.95$ ), which was acetylated to give XIIb, m.p. 204.5-206 $^{\circ}$ ;  $[\alpha]_D = +27^{\circ}$  (CHCl<sub>3</sub>);  $\lambda_{max} 306, 345 \mu$  ( $lg \epsilon = 4.21; 4.02$ ). Upon oxidation of XI with Jones reagent we obtained estra-1,3,5(10)-triene-3,17 $\beta$ -diol-3,17-diacetate-2-carboxylic acid (XIII), m.p. 183-185 $^{\circ}$ ;  $[\alpha]_D = +54^{\circ}$ ;  $\lambda_{max} 237 \mu$  ( $lg \epsilon = 3.88$ ).



By reaction of X with dimethyl sulfate we obtained 2-formyl-estra-1,3,5(10)-triene-3-methoxy-17 $\beta$ -ol (XIV), m.p. 190-192°,  $[\alpha]_D = +105^\circ$  (CHCl<sub>3</sub>).

By reaction of Va or Vb with DDQ in dioxane solution at room temperature or in benzene solution under reflux, we obtained [2,3-d]-isoxazole-estra-1,3,5(10)-triene-17 $\beta$ -ol (VIa), m.p. 178-180°,  $\lambda_{\max} 246, 288 \text{ m}\mu$  ( $\lg \epsilon = 3.72; 3.85$ ) and/or its acetate (VIb), m.p. 145-147°;  $\lambda_{\max} 246, 253, 288 \text{ m}\mu$  ( $\lg \epsilon = 3.74; 3.72; 4.00$ ).

Successively, heating VIa and VIb under reflux with a 1% solution of potassium hydroxide in methanol-water, we obtained VIIa (1).

The benzoisoxazole VIa, obtained as above described, was heated under reflux for 48 hours in dioxane solution and it appeared evident, that, under such conditions, the cleavage of the heterocyclic ring occurred, to give 2-cyano-estradiol (VIIa); this behaviour rationalizes our inability to isolate VIa in the course of the aromatization of Va with ethyl maleate in the presence of Pd/C, working in boiling dioxane.

Further aromatization experiments of 19-nor-steroids with DDQ and interpretation of the reaction mechanism will be soon reported elsewhere.

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